

# A numerical analysis of the impact of gas pressure and dielectric material on the generation of body force in an air gas plasma actuator

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## HIGHLIGHTS

- Impact of dielectric materials on plasma characteristics and body force in an SDBD plasma actuator.
- The rate of species generation in the plasma actuator is significantly influenced by the dielectric substance.
- Changing the dielectric material does not make a significant difference in electron temperature.
- Electron density changes in response to the change in the dielectric material.
- Ion density is affected by the dielectric material, with mica having the highest ion density and PTFE having the lowest.

## ABSTRACT

Plasma technology has undeniably revolutionized industrial processes in recent decades. Atmospheric pressure plasma (APP) has emerged as a prominent and widely applicable tool in various scientific disciplines. Notably, plasma-assisted flow control has become a subject of intense interest, particularly applying surface dielectric barrier discharge (SDBD) plasma actuators for aerodynamic flow control. In this study, a two-dimensional model of the SDBD plasma actuator is developed using the COMSOL Multiphysics program, incorporating air gas discharge reactions with N<sub>2</sub>/O<sub>2</sub>/Ar gases in specific ratios (0.78, 0.21, 0.01). The investigation focuses on the impact of dielectric materials (mica, silica glass, quartz, and polytetrafluoroethylene (PTFE)) on plasma characteristics and body force within the plasma actuator under constant input parameters. Moreover, the study explores how variable pressure (760, 660, and 560 torr) in different applications influences plasma properties, ultimately affecting the magnitude of the body force in the plasma actuator. These findings contribute to optimizing plasma technology for flow control applications and enhance industrial efficiency and performance.

## KEYWORDS

Dielectric Barrier Discharge  
Argon Plasma  
Plasma Actuator  
Body Force  
COMSOL Multiphysics  
Gas Discharge

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## 1 Introduction

Since nonthermal plasma first attracted scientific and technical attention 70 years ago, plasma technology has taken center stage in industrial processes both now and in the future. Over time, plasma technology developed and is today employed in everything from everyday items to cutting-edge applications (Keidar and Beilis, 2013). Various scientific sectors have utilized APP because of its outstanding quality (Mehrabifard et al., 2020; Venezia et al., 2008; Da Ponte et al., 2011; Assadi et al., 2022; Mehrabifard et al., 2017). Over the past few years, plasma-assisted

flow control has received much interest (Neretti et al., 2014; Touchard, 2008). Significant interest is in utilizing surface dielectric barrier discharge (SDBD) plasma actuators to control aerodynamic flow. Plasma actuators find utility in the realm of active airfoil leading edge separation control (Neretti et al., 2014), used for high lift (Little et al., 2010), boundary layer flow control (Porter et al., 2007; Szulga et al., 2015), handling dynamic stall in an airfoil (Post and Corke, 2006), bluff body flow control (Do et al., 2007), regulation of airflow (Neretti et al., 2012), lowering noise levels (Thomas et al., 2008), and postponing turbine blade separation (Huang et al., 2006). Many studies have

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been conducted experimentally and numerically, focusing on optimizing the ionic wind velocity and the volumetric force generation mechanism (Likhanskii et al., 2008; Jayaraman and Shyy, 2008; Mahdavi and Sohbatzadeh, 2019; Moreau, 2007). Moreau et al. reviewed the mechanical and electrical characteristics and their applications in aerodynamic flow control (Benard and Moreau, 2014). Numerous earlier numerical research has utilized two or more straightforward reactions to reduce the computations required (Abdollahzadeh et al., 2012; Boeuf et al., 2007; Likhanskii et al., 2007). Additionally, the plasma component may occasionally be viewed under an electrostatic condition (Abdelraouf et al., 2020; Kazemi et al., 2021; Omid and Mazaheri, 2020; Tehrani et al., 2022; Yu et al., 2023). In some cases, a specific sort of gas has also been applied to this structure (Mehrabifard, 2023).

This study describes a two-dimensional model of the SDBD plasma actuator. The COMSOL Multiphysics program is used in the development of the model. Air gas discharge reactions with the combination of nitrogen, oxygen, and argon gas with a ratio of 0.78, 0.21, and 0.01, respectively, are considered for this simulation. With constant input parameters, we investigate the effect of dielectric material on the plasma characteristics and body force in the plasma actuator. Besides that, the pressure can be variable in many of the mentioned applications, which can change many parameters of the plasma, which, as a result, changes the magnitude of the body force in the plasma actuator.

## 2 Model Description

### 2.1 Governing Equations

In this investigation, the fluid model was employed. The electron density and energy can be calculated by resolving the drift-diffusion formulas. To formulate the governing equations of electric discharge, the drift-diffusion approximation was adopted (Mehrabifard, 2023):

$$\frac{\partial n_e}{t} + \nabla \cdot \vec{\Gamma}_e = R_e - (\vec{u} \cdot \nabla) n_e \quad (1)$$

$$\vec{\Gamma}_e = -(\vec{\mu}_e \cdot \vec{E}) n_e - \vec{D}_e \cdot \nabla n_e \quad (2)$$

The electron continuity equation is defined by Eq. (1), where  $n_e$  is the electron density,  $D_e$  is diffusion coefficient,  $\Gamma_e$  is electron flux,  $u$  is average species velocity, and  $R_e$  is electron generation rate. Equation (2) represents the electron flow, divided into drift and diffusion. The electron energy density can be calculated using this equation:

$$\begin{aligned} \frac{\partial n_\varepsilon}{t} + \nabla \cdot \vec{\Gamma}_\varepsilon + \vec{E} \cdot \vec{\Gamma}_\varepsilon &= R_\varepsilon - (\vec{u} \cdot \nabla) n_\varepsilon \\ \vec{\Gamma}_\varepsilon &= -(\vec{\mu}_\varepsilon \cdot \vec{E}) n_\varepsilon - \vec{D}_\varepsilon \cdot \nabla n_\varepsilon \end{aligned} \quad (3)$$

The value of  $\vec{E} \cdot \vec{\Gamma}_\varepsilon$  is the amount of energy that can be extracted from an electron by applying an electric field. The following equation may be used to compute the energy gained by non-elastic collisions, which is denoted by the variable  $R_\varepsilon$ :

$$R_\varepsilon = S_{en} + \frac{Q + Q_{gen}}{q} \quad (4)$$

$S_{en}$  is the power dissipation,  $Q_{gen}$  is the heat source, and  $q$  is the electron charge.  $D_e$  is electron diffusion coefficient,  $\mu_e$  indicates energy mobility, and  $D_\varepsilon$  is energy distribution coefficient. The link between these parameters is shown in Eq. (5):

$$\begin{aligned} D_\varepsilon &= \mu_e T \\ D_e &= \mu_e T_e \\ \mu_\varepsilon &= \frac{5}{3} \mu_e \end{aligned} \quad (5)$$

The Townsend coefficients of the electron source, which are determined by the following equation, were used:

$$R_e = \sum_{j=1}^M x_j a_j N_n |\Gamma_e| \quad (6)$$

where  $M$  is the total number of reactions,  $x_j$  the molar fraction of the target species for reaction  $j$ ,  $a_j$  is the Townsend coefficient, and  $N_n$  is the total number of neutral particles are present. Considering the number  $p$  of non-elastic electron collisions, we will have:

$$R_\varepsilon = \sum_{j=1}^p x_j a_j N_n |\Gamma_e| \Delta \varepsilon_j \quad (7)$$

where  $\Delta \varepsilon_j$  is the energy dissipation of the  $j$  reaction. For non-electron-induced species, the below equation is used for mass fraction calculation:

$$\rho \frac{\partial w_k}{t} + \rho (\vec{u} \cdot \nabla) w_k = \nabla \cdot \vec{j}_k + R_k \quad (8)$$

in which  $w_k$  is the ionic density and  $j_k$  is the energy flux of the ions. The following equation obtains the electrostatic field:

$$\nabla \cdot (\varepsilon_0 \varepsilon_r E) = \rho \quad (9)$$

where  $\varepsilon_0$  is the permittivity of vacuum and  $\varepsilon_r$  is a relative dielectric constant. The following relationships are found regarding the boundary conditions for the electron flux and energy flow:

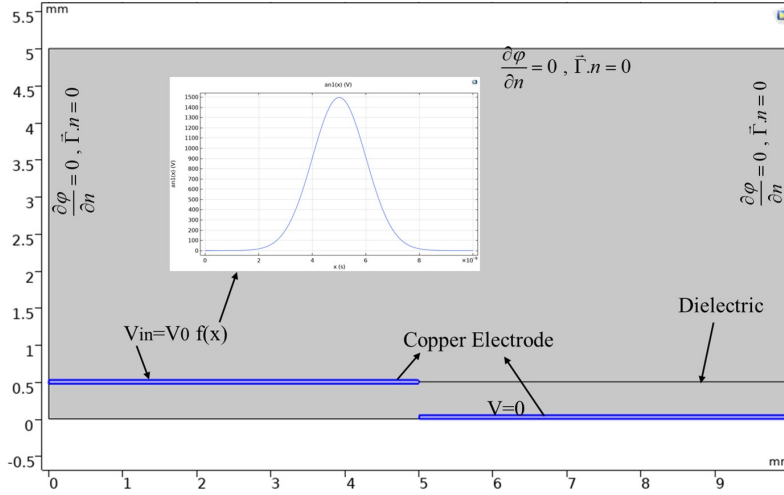
$$-\hat{n} \cdot \vec{\Gamma}_e = \left( \frac{1}{2} v_{eth} n_e \right) - \sum_p \gamma_p (\vec{\Gamma}_p \cdot \hat{n}) \quad (10)$$

$$-\hat{n} \cdot \vec{\Gamma}_\varepsilon = \left( \frac{5}{6} v_{eth} n_\varepsilon \right) - \sum_p \varepsilon_p \gamma_p (\vec{\Gamma}_p \cdot \hat{n}) \quad (11)$$

The right-hand side of Eq. (10) displays the electron number caused by the secondary electron and  $\gamma$  represents the secondary electron coefficient. Ions and excited species on the surface of electrodes are neutralized via the surface reaction. Surface interactions on the electrode are indicated by the  $\beta_j$  coefficient, which means the probability of the function of the  $j$  species. The definition of flux matching for each heavy species is as follows:

$$n \cdot j_k = M_k R_{surf,k} + M_k c_k \mu_{m,k} z_k (n \cdot E) [(z_k n \cdot E) > 0] \quad (12)$$

in which  $j_k$  and  $R_{surf,k}$  indicate the diffusive flux vector and the surface reaction rate expression for species  $k$ .  $M_k$  is mass fraction and  $c_k$  is particle mass density.



**Figure 1:** Schematic of the plasma actuator, boundary condition, and input voltage with its function.

The body force generated by plasma is:

$$f_{coulomb} = \rho_\nu \times normE \quad (13)$$

In this equation,  $\rho_\nu$  is the density of electrons and positive/negative ions,  $normE$  is the normalized electric field in the  $x - y$  direction. In this simulation, the value of charge density is equal to:

$$\rho_\nu = (n_{Ar^+} + n_{N_{2^+}} + n_{O_{2^+}} + n_{N_{4^+}} + n_{O_{4^+}} - n_e - n_{O^-}) \quad (14)$$

**Table 1:** The reactions of electron impact with active species of nitrogen. (Singly ionized nitrogen molecule  $N_2^+$ , metastable nitrogen  $N_{2s}$ , nitrogen molecule  $N_2$ ) (Mehrabiard, 2023; Hagelaar, 2010; LXcat, 2023)

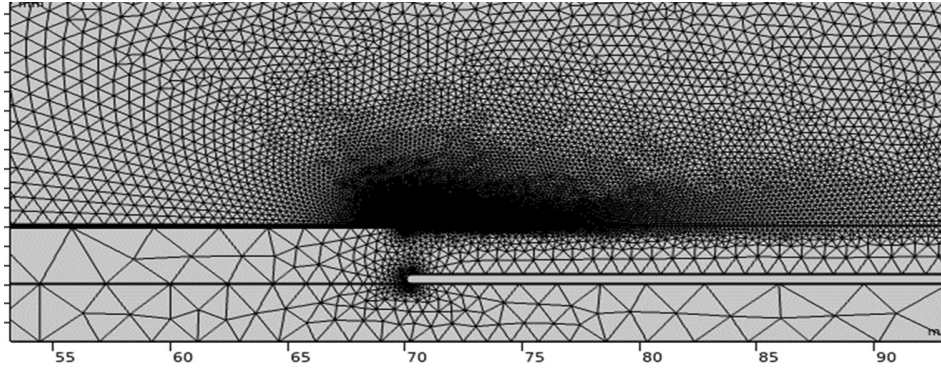
Reactions	Formula	Type	$\Delta\varepsilon$ (eV)
1	$e + N_2 \rightarrow e + N_2$	Elastic	0
2	$e + N_2 \rightarrow e + N_{2s}$	Excitation	0.02
3	$e + N_2 \rightarrow e + N_{2s}$	Excitation	0.29
4	$e + N_2 \rightarrow e + N_{2s}$	Excitation	0.291
5	$e + N_2 \rightarrow e + N_{2s}$	Excitation	0.59
6	$e + N_2 \rightarrow e + N_{2s}$	Excitation	0.88
7	$e + N_2 \rightarrow e + N_{2s}$	Excitation	1.17
8	$e + N_2 \rightarrow e + N_{2s}$	Excitation	1.47
9	$e + N_2 \rightarrow e + N_{2s}$	Excitation	1.76
10	$e + N_2 \rightarrow e + N_{2s}$	Excitation	2.06
11	$e + N_2 \rightarrow e + N_{2s}$	Excitation	2.35
12	$e + N_2 \rightarrow e + N_{2s}$	Excitation	6.17
13	$e + N_2 \rightarrow e + N_{2s}$	Excitation	7
14	$e + N_2 \rightarrow e + N_{2s}$	Excitation	7.35
15	$e + N_2 \rightarrow e + N_{2s}$	Excitation	7.36
16	$e + N_2 \rightarrow e + N_{2s}$	Excitation	7.8
17	$e + N_2 \rightarrow e + N_{2s}$	Excitation	8.4
18	$e + N_2 \rightarrow e + N_{2s}$	Excitation	8.16
19	$e + N_2 \rightarrow e + N_{2s}$	Excitation	8.55
20	$e + N_2 \rightarrow e + N_{2s}$	Excitation	8.89
21	$e + N_2 \rightarrow e + N_{2s}$	Excitation	11.03
22	$e + N_2 \rightarrow e + N_{2s}$	Excitation	11.88
23	$e + N_2 \rightarrow e + N_{2s}$	Excitation	12.25
24	$e + N_2 \rightarrow e + N_{2s}$	Excitation	13
25	$e + N_2 \rightarrow 2e + N_2^+$	Ionization	15.6

**Table 2:** The interactions between oxygen and electrons (Hagelaar, 2010; LXcat, 2023)

Reactions	Formula	Type	$\Delta\varepsilon$ (eV)
1	$e + O_2 \rightarrow e + O_2$	Elastic	-
2	$e + O_2 \rightarrow O + O^-s$	Attachment	-
3	$e + O_2 \rightarrow e + O_2$	Excitation	0.02
4	$e + O_2 \rightarrow e + O_2$	Excitation	0.19
5	$e + O_2 \rightarrow e + O_2$	Excitation	0.19
6	$e + O_2 \rightarrow e + O_2$	Excitation	0.38
7	$e + O_2 \rightarrow e + O_2$	Excitation	0.38
8	$e + O_2 \rightarrow e + O_2$	Excitation	0.57
9	$e + O_2 \rightarrow e + O_2$	Excitation	0.75
10	$e + O_2 \rightarrow e + O_2 a1d$	Excitation	0.977
11	$e + O_2 a1d \rightarrow e + O_2$	Excitation	-0.977
12	$e + O_2 \rightarrow e + O_2 b1s$	Excitation	1.627
13	$e + O_2 b1s \rightarrow e + O_2$	Excitation	-1.627
14	$e + O_2 \rightarrow e + O_2 45$	Excitation	4.5
15	$e + O_2 45 \rightarrow e + O_2$	Excitation	-4.5
16	$e + O_2 \rightarrow e + O + O$	Dissociation	6
17	$e + O_2 \rightarrow e + O + O1d$	Excitation	8.4
18	$e + O_2 \rightarrow e + O1s$	Excitation	9.95
19	$e + O_2 \rightarrow 2e + O_2^+$	Ionization	12.06

## 2.2 Boundary Conditions and Reactions

An asymmetric pair of copper electrodes separated by a dielectric substance from the actuator. While the other electrode is connected to the ground and encased by the dielectric material, the first electrode is placed on the dielectric surface and in contact with the gas flow at atmospheric pressure. A Gaussian voltage power supply drives the discharge at 1.5 kV. The electron density is first estimated to be  $10^8 \text{ m}^{-3}$ . The gas is at a temperature of 293.15 K and a pressure of 760 torr. In the simulation, the material components are defined as a model of the air, considering the related reactions of the species. Rate coefficients were obtained by solving Boltzmann's equation with BOLSIG+ (Hagelaar and Pitchford, 2005) and the cross-sections from the LXCAT data source (Pitchford et al., 2017). Reaction rates were taken from the references (Sakiyama et al., 2012). The dominant reactions can be seen in Tables 1 to 6, including electron impact



**Figure 2:** Meshing and its density in edges.

**Table 3:** Atomic and molecule-to-molecule interactions with two and three bodies (Stafford and Kushner, 2004; Sohbatzadeh and Soltani, 2018).

Reactions	Formula	Type	$K^f$ ( $\text{m}^3 \cdot \text{s}^{-1} \cdot \text{mol}^{-1}$ )
1	$e + \text{N}^+ \rightarrow \text{N}$	Recombination	$3.5 \times 10^{-18}$
2	$e + \text{N}_2 \rightarrow 2e + \text{N} + \text{N}^+$	Dissociative ionization	$2.4 \times 10^{-23}$
3	$e + \text{N}_2 \rightarrow e + 2\text{N}$	Dissociative	$2 \times 10^{-17}$
4	$e + \text{N}_2^+ \rightarrow 2\text{N}$	-	$2.8 \times 10^{-13}$
5	$\text{N}^+ + \text{N}_2 \rightarrow \text{N} + \text{N}_2^+$	Charge exchange	$10^{-17}$

**Table 4:** Atomic and molecule-to-molecule interactions with two and three bodies (Stafford and Kushner, 2004; Sohbatzadeh and Soltani, 2018).

Reactions	Formula	$K^f$ ( $\text{m}^3 \cdot \text{s}^{-1} \cdot \text{mol}^{-1}$ )
1	$\text{O} + \text{O}_2 + \text{O}_2 \rightarrow \text{O}_3 + \text{O}_2$	$6 \times 10^{-46} \times (1.3^{-2.8})$
2	$\text{O} + \text{O}_2 + \text{N}_2 \rightarrow \text{O}_3 + \text{N}_2$	$5.6 \times 10^{-46} \times (1.3^{-2.8})$
3	$\text{O} + \text{O}_3 \rightarrow \text{O}_2 + \text{O}_2$	$8 \times 10^{-18} \times \exp(-2060/4)$
4	$\text{O} + \text{NO}_2 \rightarrow \text{NO} + \text{O}_2$	$5.6 \times 10^{-7} \times \exp(180/40)$
5	$\text{O} + \text{N O}_3 \rightarrow \text{O}_2 + \text{NO}_2$	$1.7 \times 10^{-17}$
6	$\text{O} + \text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_2 + \text{O}_2$	$1 \times 10^{-22}$
7	$\text{N} + \text{O}_2 \rightarrow \text{NO} + \text{O}$	$1.5 \times 10^{-7} \times \exp(8)$
8	$\text{N} + \text{O}_3 \rightarrow \text{NO} + \text{O}_2$	$1 \times 10^{-22}$
9	$\text{N} + \text{NO} \rightarrow \text{N}_2 + \text{O}$	$2.1 \times 10^{-11} \times \exp(0.25)$
10	$\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$	$3 \times 10^{-18} \times \exp(-3.8)$
11	$\text{N} + \text{NO}_2 \rightarrow \text{N}_2\text{O} + \text{O}$	$5.8 \times 10^{-18} \times \exp(0.55)$
12	$\text{NO}_2 + \text{O}_3 \rightarrow \text{NO}_3 + \text{O}_2$	$1.4 \times 10^{-19} \times \exp(-6.2)$
13	$\text{O}_2\text{b1s} + \text{O}_2 \rightarrow \text{O}_3 + \text{O}$	$4.8 \times 10^{-21}$
14	$\text{N}_2 + \text{O}_2 \rightarrow \text{N}_2\text{O} + \text{O}$	$6 \times 10^{-20} \times (1.3^{0.55})$
15	$\text{O}^- + \text{O} \rightarrow \text{O}_2 + e$	$2 \times 10^{-16} \times (400^{0.5})$
16	$\text{O}^- + \text{O}_2 \rightarrow \text{O}_3 + e$	$3 \times 10^{-16} \times (400^{0.5})$
17	$\text{O}^- + \text{O}_2 \rightarrow \text{O} + \text{O}_2 + e$	$6.9 \times 10^{-16} \times (400^{0.5})$
18	$\text{O}_2 + \text{O}_2 \rightarrow \text{O}_3 + \text{O}$	$2.95 \times 10^{-27} \times (400^{0.5})$
19	$\text{O}^- + \text{O}_3 \rightarrow \text{O}_2 + \text{O}_2 + e$	$3 \times 10^{-16} \times (400^{0.5})$
20	$\text{O}^- + \text{O}_3 \rightarrow \text{O}_2^- + \text{O}_2$	$1.02 \times 10^{-17} \times (400^{0.5})$
21	$\text{O}_2^- + \text{O} \rightarrow \text{O}_3 + e$	$1.5 \times 10^{-16} \times (400^{0.5})$
22	$\text{O} + \text{O}_3 \rightarrow \text{O}_2 + \text{O} + \text{O}$	$1.2 \times 10^{-16}$
23	$\text{O}_2 \rightarrow \text{O}_2$	0.2
24	$\text{O}_2 \rightarrow \text{O}$	$10^{-5}$
25	$\text{N}^+ + \text{O}_2 \rightarrow \text{N} + \text{O}_2^+$	$3 \times 10^{-16}$

ionization, electron attachment, elastic collisions, excitation, recombination, neutral component collisions, and ion conversion processes. The schematic structure of SDBD and its boundary conditions is shown in Fig. 1. Copper electrode dimensions are 0.05 mm  $\times$  5 mm, and the dielectric dimensions are 0.5 mm  $\times$  10 mm. We employ five boundary layers and a 1.4 stretching factor for the system's whole boundary while meshing this structure. A

free triangular mesh with a maximum element size of 0.4 is employed for the entire geometry (Fig. 2).

Nitrogen, oxygen, and argon reactions with a mass fraction of 0.78, 0.21, and 0.01 are used to simulate air gas discharge. The electron impact reactions with active nitrogen species, such as nitrogen molecules  $\text{N}_2$  and  $\text{N}_2^+$ , that are singly ionized are shown in Table 1. The interactions of electron impact with active oxygen species, such

**Table 5:** Atomic and molecule-to-molecule interactions with two and three bodies (Sohbatzadeh and Soltani, 2018).

Reactions	Formula	$K^f$ ( $\text{m}^3 \cdot \text{s}^{-1} \cdot \text{mol}^{-1}$ )
1	$\text{N}_2 + \text{Ar} + \text{N}_2^+ \rightarrow \text{Ar} + \text{N}_4^+$	$1.8 \times 10^6$
2	$\text{e} + \text{N}_4^+ \rightarrow \text{N}_2 + \text{N}_2$	$1.2 \times 10^{11}$
3	$\text{O}_2 + \text{Ar} + \text{O}_2^+ \rightarrow \text{Ar} + \text{O}_a^+$	$2 \times 10^5$
4	$\text{O}_4^+ + \text{Ar} + \text{O}^- \rightarrow 2\text{O}_2 + \text{Ar} + \text{O}$	$5.2 \times 10^{11}$
5	$\text{Ar} + \text{O}_4^+ \rightarrow \text{Ar} + \text{O} + \text{O}_2 + \text{e} + \text{O}^+$	$6 \times 10^7$
6	$\text{O}^- + \text{Ar} + \text{O}_4^+ \rightarrow 2\text{O}_2 + \text{O} + \text{Ar}$	$3.8 \times 10^8$
7	$\text{N}_2 + \text{Ar} + \text{N}_2^+ \rightarrow \text{Ar} + \text{N}_4^+$	$1.8 \times 10^6$

**Table 6:** The interactions between electrons and Argon (Hagelaar, 2010; LXcat, 2023).

Reactions	Formula	Type	$\Delta\varepsilon$ (eV)
1	$\text{e} + \text{Ar} \rightarrow \text{e} + \text{Ar}$	Elastic	0
2	$\text{e} + \text{Ar} \rightarrow \text{e} + \text{Ar}_s$	Excitation	11.5
3	$\text{e} + \text{Ar}_s \rightarrow \text{e} + \text{Ar}$	Superelastic	-11.5
4	$\text{e} + \text{Ar} \rightarrow 2\text{e} + \text{Ar}^+$	Ionization	15.8
5	$\text{e} + \text{Ar}_s \rightarrow 2\text{e} + \text{Ar}^+$	Ionization	4.24
6	$\text{Ar}_s + \text{Ar}_s \rightarrow \text{e} + \text{Ar} + \text{Ar}^+$	Penning ionization	-
7	$\text{Ar}_s + \text{Ar} \rightarrow \text{eAr} + \text{Ar}$	Metastable quenching	-

as metastable oxygen  $\text{O}_2^s$ , oxygen molecule  $\text{O}_2$ , and singly ionized oxygen molecule  $\text{O}_2^+$ , are shown in Table 2. Tables 3, 4, and 5 offer the two and three-body reaction rates between atoms and molecules. Table 6 shows all electron impact reactions with argon and its molecule-to-molecule reactions. Fifteen surface reactions are considered in addition to the aforementioned reactions, as indicated in Table 7.

### 3 Results and discussion

An SDBD plasma actuator's body force magnitude has been obtained from the simulations. The rate of species generation is significantly influenced by the dielectric substance. Dielectric materials are electrically insulating substances with low electrical conductivity. Numerous processes that alter the behavior of the plasma and the emergence of species may take place when it comes into contact with a dielectric substance (Fridman and Kennedy, 2016; Lieberman and Lichtenberg, 1994).

**Table 7:** Table of surface reactions in air discharge (Sohbatzadeh and Soltani, 2018).

Reactions	Formula	TSticking Coefficient
1	$\text{O}_2\text{ald} \rightarrow \text{O}_2$	1
2	$\text{O}_245 \rightarrow \text{O}_2$	1
3	$\text{O}_2\text{bls} \rightarrow \text{O}_2$	1
4	$\text{O}_2 \rightarrow \text{O}_2$	1
5	$\text{O}_2^+ \rightarrow \text{O}_2$	1
6	$\text{O}^+ \rightarrow \text{O}$	1
7	$\text{O}^- \rightarrow \text{O}$	1
8	$\text{O1s} \rightarrow \text{O}$	1
9	$\text{O1d} \rightarrow \text{O}$	1
10	$\text{N}^+ \rightarrow \text{N}$	1
11	$\text{Ar}_s \rightarrow \text{Ar}$	1
12	$\text{Ar}^+ \rightarrow \text{Ar}$	1
13	$\text{N}_2^s \rightarrow \text{N}_2$	1
14	$\text{N}_2^+ \rightarrow \text{N}_2$	1

In this study, we investigated how four different dielectric materials affected plasma properties that change the magnitude of body force in the plasma actuator. Figure 3 displays the electrical potential when plasma is formed. The grounded electrode has zero potential, while the power electrode has a voltage of 1.5 kV with a Gaussian shape.

Electron density and temperature play a vital role in plasma and are the initiators of many plasma reactions. And in many applications, the effect of electron density has been investigated (Mehrabifard et al., 2023; Tanaka et al., 2015), and it is an effective factor for the plasma actuators. The evaluation of electron temperature at 6 ns for various dielectric materials is shown in Fig. 4. Moving away from the power electrode causes the electron temperature to decrease from its highest value. As the figure shows, the temperature changes for each dielectric material were almost in the same range. Changing materials does not make a significant difference in electron temperature. The temperature of the electron is directly influenced by the electric field due to the implementation of a constant potential function, resulting in minimal fluctuations in temperature.

Then, once more, Eqs. (12) and (13) say this: the magnitude of the body force is directly influenced by the particle density. When all other factors remain the same, Fig. 5 illustrates how much the electron density changes with respect to the dielectric material. Using mica as dielectric results in electrons in its maximum value; quartz and silica have almost the same amount,  $8.61 \times 10^{14} \text{ m}^{-3}$  and  $8.22 \times 10^{14} \text{ m}^{-3}$ , respectively, and PTFE has the lowest value of  $2.39 \times 10^{14} \text{ m}^{-3}$ . Changes in electron density are influenced by many causes, including surface charge accumulation and the photoionization effect. Among these factors, the alteration of the dielectric coefficient is particularly significant in determining the changes in surface charge accumulation. Indeed, the density rises in proportion to the increase in the dielectric coefficient.

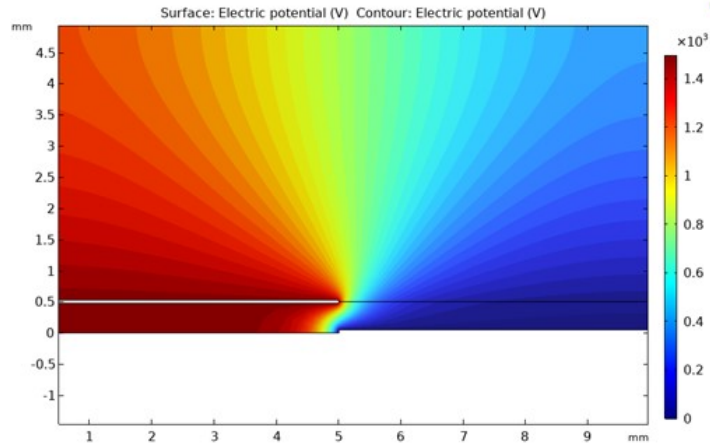


Figure 3: Distribution of electric potential for  $V_{in} = 1.5$  kV.

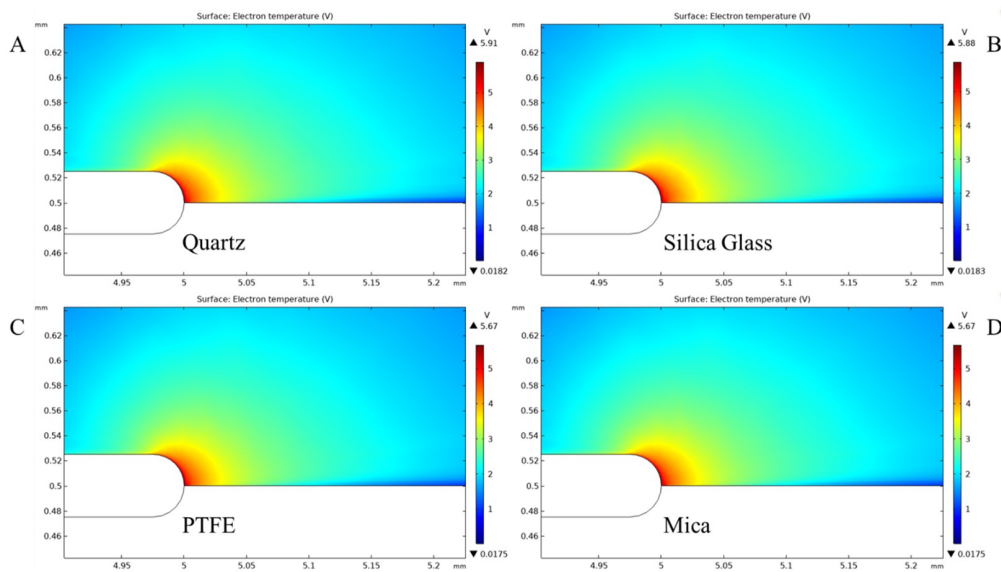


Figure 4: The electron temperature over 6 ns at 1.5 kV with Gaussian function for different dielectric materials (A) Quartz, (B) Silica Glass, (C) PTFE, (D) Mica.

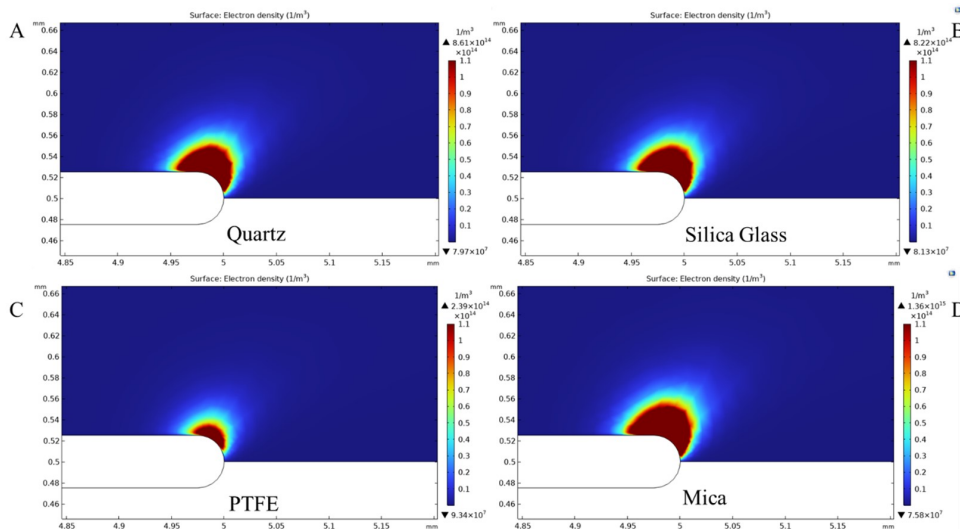


Figure 5: The electron densities over 6 ns at 1.5kV with Gaussian shape for different dielectric materials (A) Quartz, (B) Silica Glass, (C) PTFE, (D) Mica.

The ion densities are shown in Fig. 6. The  $n_{Ar^+}$ ,  $n_{N_2^+}$ ,  $n_{O_2^+}$ ,  $n_{N_4^+}$ ,  $n_{O^+}$ ,  $n_{O_4^+}$ , and  $n_{O^-}$  are the main species mea-

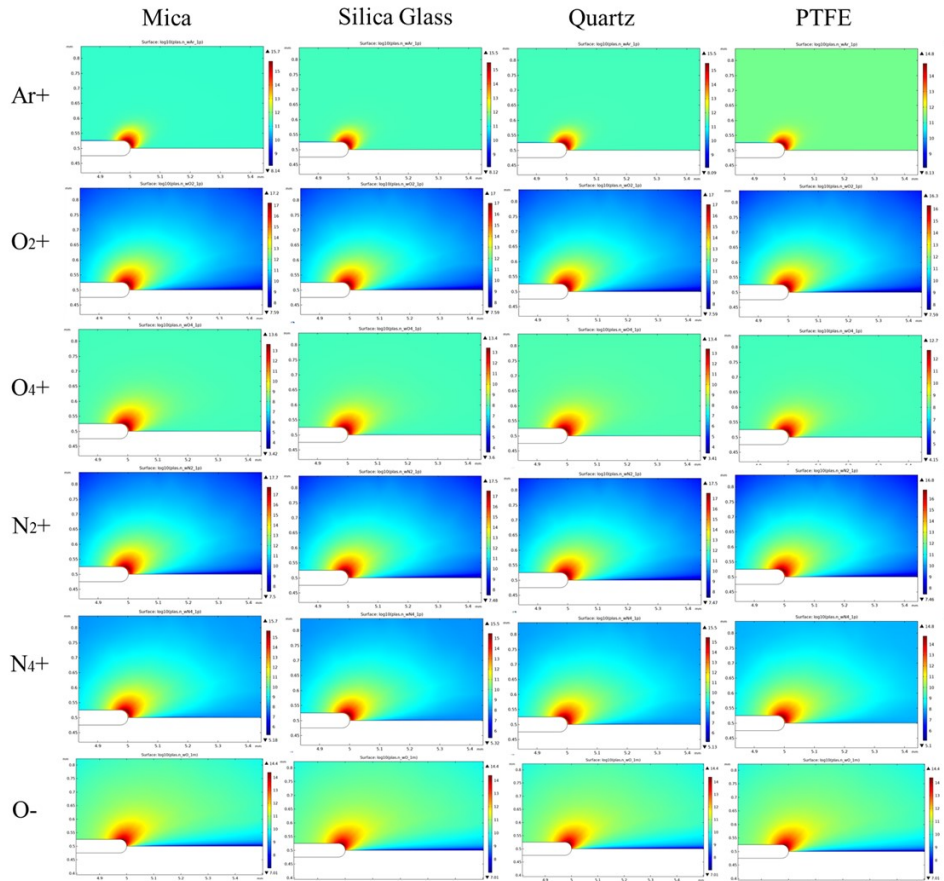


Figure 6: Two-dimensional distribution of the ion densities for different dielectric materials.

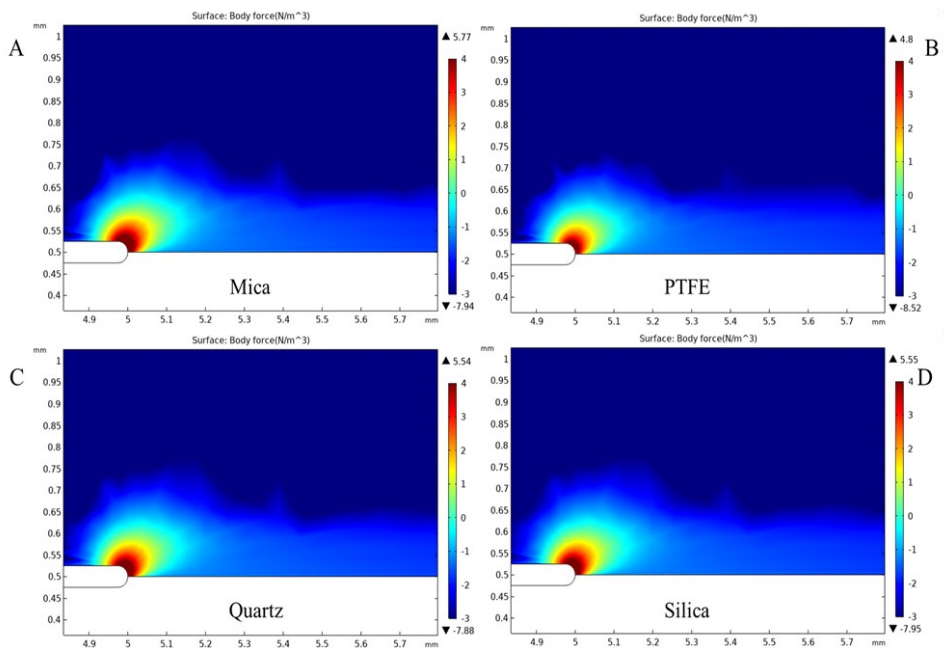
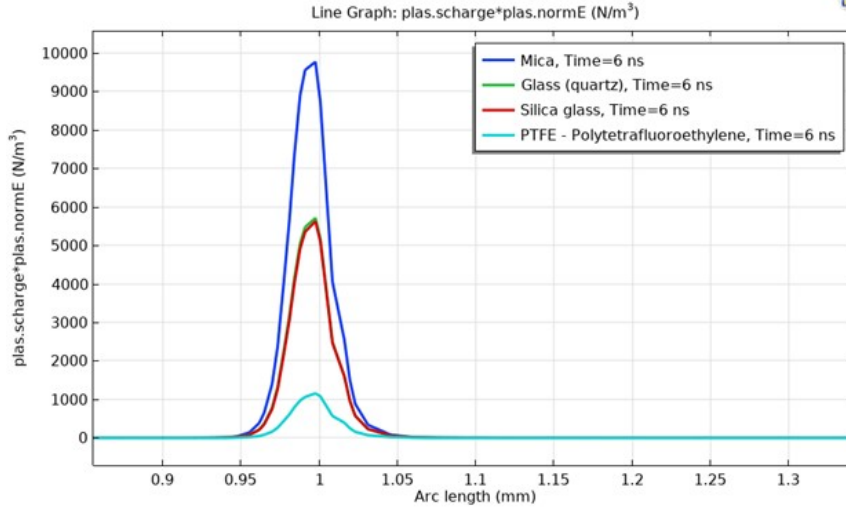


Figure 7: Magnitude of body force in the presence of different dielectric barrier discharge.

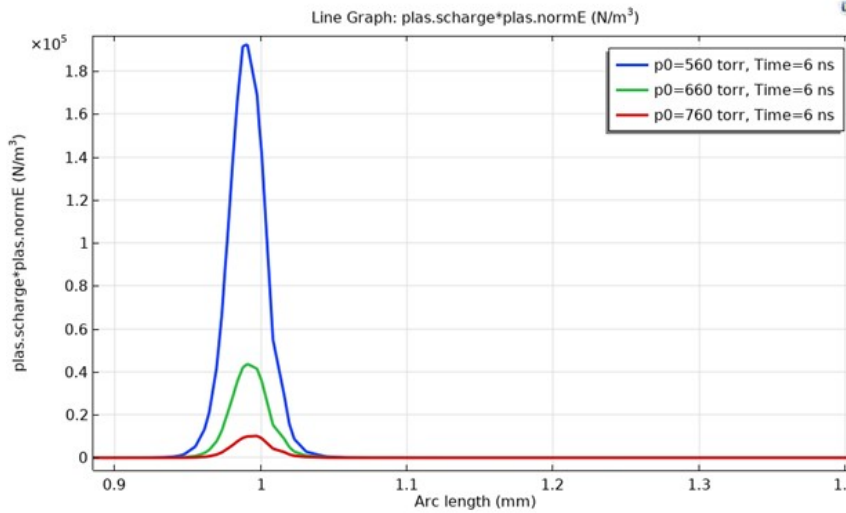
sured in this simulation in the presence of different dielectric materials. As Fig. 6 shows, all ion densities have a higher value in the presence of mica and the lowest for PTFE. And the value for quartz and silica glass shows almost the same ion density. As stated for the electron.

The increase in surface charge accumulation will affect the density of ions, which is related to the dielectric coefficient.

The body force can be calculated from the difference between positive and negative charges and the magnitude of the normal field in the discharge space. According to



**Figure 8:** The body force is created on the surface of the electrode and the dielectric.



**Figure 9:** The effect of different pressures on body force.

Eq. (13), the logarithmic body force distribution for dielectric materials is shown in Fig. 7. To investigate more precisely, a hypothetical line on the surface of the electrode and dielectric is considered. This line's beginning and end points are  $(x = 4, y = 0.54)$  and  $(x = 8, y = 0.54)$ . Figure 8 shows a non-logarithmic magnitude of the body force on this virtual line. As it is known, the force of the body is the highest for Mica, and its magnitude will be equal to  $9800 \text{ N.m}^{-3}$ . It will be equal to  $5700 \text{ N.m}^{-3}$ ,  $5600 \text{ N.m}^{-3}$ , and  $1100 \text{ N.m}^{-3}$  for quartz, silica, and PTFE, respectively.

Gas pressure is one of the main parameters that can change the characteristics of the plasma; as a result, this causes a change in the magnitude of the body force in the plasma actuators. The pressure of 560, 660, and 760 torr are considered for plasma simulation. Figure 9 shows changes in body force magnitude for different pressures. The body force is measured on a virtual line on the upper part of the power electrode. As shown in the figure, the magnitude of body force changes dramatically by reducing pressure. The pressure was reduced by 100 torr in each

stage, but significant changes were shown in body force.

Changing the dielectric coefficient, the electron energy distribution function changes around the power electrode, and this change leads to different ionization collisions around this electrode, which ultimately creates a difference in the ionization coefficient and volumetric force. As the ambient pressure changes, the ratio of the electric field to the ambient pressure will change, considering that the drift velocity is a function of the electric field-to-pressure ratio ( $E/P$ ). On the other hand, this velocity is directly related to electron mobility; it can affect the changes in electron mobility. In fact, the set of these net charge density changes that lead to the production of propulsion force will be different.

## 4 Conclusions

Active current control is one of the areas where plasma is used. Additionally, modeling these systems before construction may save time and money and ensure the cre-



ation of an efficient system. From the outcome of our investigation, it is possible to conclude that without changing the main parameters and only by changing the type of dielectric material, the magnitude of the body force can be increased. Among the selected materials, using mica, we have the most body force. Furthermore, it is worth noting that pressure, being a crucial factor in plasma production, has shown that even the slightest alteration may lead to substantial variations in the body's force. Moreover, due to the significant change in body force, pressure is one of the parameters that should be considered in the actuator design. The simulation findings, performed with consideration of air gas discharge, can be valuable in the design of plasma drive systems for many applications, enabling the attainment of optimum outcomes via the selection of appropriate materials.

## Conflict of Interest

The authors declare no potential conflict of interest regarding the publication of this work.

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